NITRATE-NITRITE, VITAMIN C AND IN-VITRO METHEMOGLOBIN FORMATION
FROM SOME VEGETABLES

by

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ABSTRACT

The content of nitrate, nitrite and Vitamin C in eleven samples of fresh vegetables were determined. The levels of nitrate were beyond more partial pa

When an in vitro test of methemoglobin formation was carried out with extract from the samples, modified hemoglobin formed paralleled content of nitrate in the vegetables. It is suggested that caution must be exercised in the use of some of these vegetables in infant foods as they may contribute to induction of methemoglobin.

INTRODUCTION

The hazards associated with nitrate-nitrite in the environment have been reviewed (1 - 4). Although nitrate as such does not constitute a bazard, there exists the possibility of conversion of nitrates to hazard, there exists the possibility of conversion of carcinogenic nitrites (5) which can in turn serve as precursors of carcinogenic nitrosamines (6). The main hazard associated with nitrate in foods is nethemoglobinemia in infants (7).

Nitrates and nitrites may occur naturally in high concentrations in regetables consumed by humans or in forages fed to livestock. Intake of nitrate also comes from water supplies that are high in nitrate content (1). Most vegetables contain a large proportion of indigestible fibrous aterials but they are also good sources of minerals, vitamins and roteins. Leafy vegetables are generally used by Nigerians in the reparation of soup, therefore, the leaf protein content of most of them has been determined (8). Some leaf proteins have also been used in the nutritional rehabilitation of malnourished children (9).

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Recently, the nitrate-nitrite in some cereals and baby foods used in Nigeria was reported (10). However, such studies have not been done on vegetables used during weaning and rehabilitation of malnourished children objective of this study is to determine the nitrate-nitrite and accompanying vitamin C. which may serve in conjunction with other reducing substances to convert nitrate to nitrite, in some of these vegetables. The work is extended to the determination of the potential capacity of the

METHODS

(a) Preparation of samples.

All samples were obtained fresh from the University of Benin Farm project (a project established to grow local vegetables with modern techniques of farming and application of fertilizers) in the morning and stored in cellophane bags prior to determination on the same day.

The bulk sample in each case was divided into three parts. Each sub-sample of log was homogenized with 20ml of distilled water and later made up to 200ml in a standard flask. The flask was quickly immersed in a water bath (80°C) and held for 10 minutes. The resulting extract was passed through a column of aluminium oxide to remove the pigments and then filtered through No. 41 whatman paper for analyses.

(b) Estimation of Nitrate-nitrite and Vitamin C.

The method based on reduction of nitrate to nitrite with cadmium followed by the formation of a diazo compound with sulfanilic acid and N-1-napthyl-ethylenediamine introduced by Elliot and Porter (11) for nitrate in bacon and later modified for determination of nitrate in vegetables by Kenny and Walshe (12) was used. The nitrite due to the reduction of nitrate was calculated by substracting the determination done without reduction on some of the same sample.

The procedure for the determination of ascorbic acid (Vitamin C) was by Wahba \underline{et} al (13).

For the study of the effect of cooking on nitrate-nitrite, fresh samples of some vegetables were sub-sampled and parts were boiled in distilled water for thirty minutes, extracted and analysed for nitrate-nitrite.

(c) In vitro methemoglobin formation.

Blood was obtained from the blood bank of the University of Benin Teaching Hospital, Benin City. Preparation of Oxyhemoglobin and Methemoglobin with sodium nitrite was as reported by Russo and Sorstokke (14). To determine the potential capacity of the vegetable in forming methemoglobin, five leafy vegetables (four on basis of high nitrate content and one based on low nitrate content) were extracted, prepared and the nitrate content was reduced to nitrite with Cadmium. 3ml of Oxyhemoglobin solution with absorbance of 0.8 at 500nm was pipetted into a

cuvette followed by 3ml of 0.2M Potassium phosphate buffer (pH 7.0).

1ml thoroughly mixed. The formation of methomography and the content thorough the content of the cuvette and the content was crease in absorbance reading at 560nm at 5 minutes interval for 80 minutes in one case of the cuvette and the content of the cuvette and the The formation of methemoglobin was determined from ce reading at Frommethemoglobin was determined for 80 decrease in decrease in decreasing at 560nm at 5 minutes interval for 80 minutes in one case of the vegetable (Telfaris, spp) minutes and up to minutes in one case of the vegetable (Telfaris, spp) with low nitrate level. A control experiment was run simultaneously with minutes as now run simultaneously with with low file acontrol experiment was run simultaneously with alent concentration of sodium nitrite as per nitrite determined for vegetable formed. equivalent consequence of sodium nitrite as per nitrite determined for the moglobin formed. the moglobin formed.

RESULTS AND DISCUSSION

Table 1 shows the level of nitrate-nitrite in eleven vegetables Table or nitrate-nitrite in eleven vegetable commonly consumed in Nigeria. Celosia sp contains about twice the commonly constants to levels (300ppm) in vegetables about twice the recommended by FAO (16). Other vegetables (15), and above the 500ppm are beyond the recommended limit of 300ppm are 500ppm recommended limit of 300ppm are Lactuca sativa, which are bybridus, Brassica oleracea variety acephala, Talinum maranthus nystracea oleracea and Veronica sp. Other vegetables analysed have safe levels of nitrate, whilst Telfaris sp has only trace

Nitrate-nitrite content in vegetables Table 1:

Sample	Number of sample	Range (ppm)	Mean (ppm)	C.V.%
Celosia SP	3	500 - 673	568	31
lactuca sativa	3	467 - 553	500	. 17
Amaranthus hybridus	3	450 - 555	495	21
Brassica oleracea Variety acephala	3	460 - 534	492	16
lalinum triangulare	3	370 - 440	400	18
Spinacia oleracea	3	320 - 355	342	10
aucus carota	4	190 - 300	256	43
eronica sp		190 - 237	211	23
isca sp	3	117 - 124	120	6
rassica oleracea	3		95	1
chorus of the	. 3	90 - 107	20	2
efaris sp	3	18 - 22	. 20	
	* 4		١	

Insignificant quantities of nitrite were detected in Celosia sp (16.7ppm) and Daucus carota (12.5ppm).

Table 2 shows the vitamin C content of the vegetables examined. level of vitamin C was detected in Celosia sp, whilst relatively low levels were estimated in other vegetables with high nitrate content.

Vitamin C content in vegetables Table 2:

Sample	Number of sample	Range (mg/100g)	Mean (mg/100g)	C.V.9
<u>Celosia</u> sp Lactuca sativa .	3	130 - 176	158	30
Amaranthus hybridus	3	16 - 22 30 - 48	14	32
Brassica oleracea variety acephala	3	23 - 32	38 27	47
Talinum triangulare	3	47 - 51	49	33
Spinacia <u>oleracea</u>	3	61 - 67	65	9
Daucus carota Veronica sp	3	21 - 29 80 - 83	. 24 82	34
Brassica oleracea	3	41 - 59	48	38
Corchorus olitorus	3	69 - 74	71	7
<u>elefaris</u> sp	3	48 - 56	52	15

It is as well that, except for Celosia sp, all other vegetables with high nitrate content have low vitamin C. Storage of vegetables with high reducing substances (vitamin C is a naturally occurring reducing agent) and high level of nitrate may bring about a quantitative conversion of nitrate to nitrite. Such vegetables may then become unsafe for use in infant foods. Reductive conversion for Spinach has been reported by Phillips (17). (17). Perhaps the trace amounts of nitrite in Celosia sp and Daucus represent the product of such initial reduction.

Table 3 shows that less than a 25% of the nitrate is lost during cooking, although cooking may serve to destroy nitrate reductase present in fresh vegetables, thereby eliminating a pathway of nitrate reduction

the enzyme may become significant in nitrate conversion when

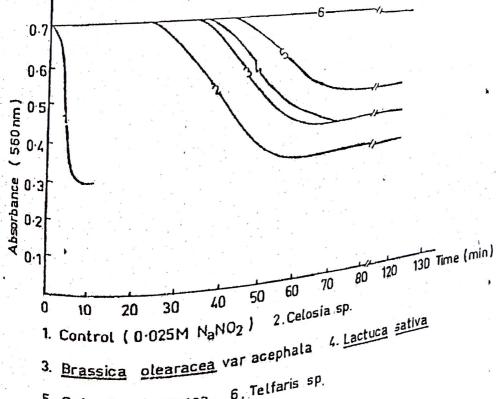
gegetable 3: Effect of cooking on Nitrate-nitrite in vegetables

		Nitrate-nitrite (ppm)		
samp1e	· · · · · · · · · · · · · · · · · · ·	Cooked	Fresh	% loss of Nitrate-nitrite
celosia sp		413	530	
Amaranthus hybridus	· . ,	466	555	21 16
veronica sp		190	237	20

boiled at 100°C for 30 minutes.

Fig. 1 and Table 4 show the kinetics of methemoglobin formation with vegetable extracts. In Fig. 1 control experiments run with 0.025M NaNo, had a change of 0.4 absorbance unit in 5 minutes, while the extract with Celosia sp had a similar change (0.37A) in thirty minutes after about an Initial thirty minutes lag period. The kinetics of the other vegetables





6. Telfaris sp. 5. Spinacia olearacea

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had similar activity curves which were all consistent with an initial step that involved a single electron transfer from nitrite to the bound dioxygen of oxyhemoglobin (19).

Table 4: Methemoglobin formation with vegetable extract

Sample	Absorbance	Time (min)	A/min x 100
Control-0.025M NaNO2	0.40	5	
Celosia sp	0.37	30	8.00 1.23
Brassica oleracea variety acephala	0.26	30	0.87
Lactuca sativa	0.25	30	0.83
Spinacia oleracea	0.20	30 1	0.67
Telfaris sp	0.02	30	0.07

The steepness of the slope, the initial lag period (figure 1) or rate of methemoglobin formation per minute (Table 4) would therefore provide a quantitative assessment of the in vitro capacity of the vegetables in forming methemoglobin when run and compared with the appropriate level regarded as unsafe. Celosia sp approached such an unsafe level, while methemoglobin was virtually unformed with extract from Telfaris sp.

This study serves to indicate that nitrate levels in some Nigerian vegetables, especially in Celosia sp, Brassica oleracea variety acephala and Lactuca sativa may warrant periodic monitoring, especially with increasing use of fertilizers in the country. Nitrite consumed from all sources in Nigeria has not been published, although a recent investigation was reported on the critical nitrate-nitrite-nitrosamine levels of some fermented Nigerian beverages (20). What is currently being sought is the reduction in the rate of nitrates in vegetables and from all other squrces through various contracts which provide for the limited use of fertilizers, the choice of varieties, preventive health measures and application of relevant technology to enable the accepted norm to be fed to infants, thus reducing the risk of methemoglobinemia (7).

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